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Adsorption-desorption noise can be used for improving selectivity

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Abstract 8

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Small chemical sensors are subjected to adsorption-desorption fluctuations which usually considered as noise contaminating useful 9 signal. Based on temporal properties of this noise, it is shown that it can be made useful if proper processed. Namely, the signal, which 10 characterizes the total amount of adsorbed analyte, should be subjected to a kind of amplitude discrimination (or level crossing discrimina-11 tion) with certain threshold. When the amount is equal or above the threshold, the result of discrimination is standard dc signal, otherwise 12 13 it is zero. Analytes are applied at low concentration: the mean adsorbed amount is below the threshold. The threshold is achieved from time to time thanking to the fluctuations. The signal after discrimination is averaged over a time window and used as the output of the whole 14 device. Selectivity of this device is compared with that of its primary adsorbing sites, based on explicit description of the threshold-crossing 15 16 statistics. It is concluded that the whole sensor may have much better selectivity than do its individual adsorbing sites. 17 © 2003 Published by Elsevier B.V.

Keywords: Sensor; Fluctuations; Noise; Adsorption; Selectivity; Electronic nose 18

1. Introduction 19

20 Detectors of chemical substances are usually based on selective adsorption-desorption (binding-releasing) of ana-21 lyzed chemicals by specific adsorbing sites (receptor 22 molecules). The receptor molecules are attached to an elec-23 tronic device able to measure the amount of the analyte 24 25 adsorbed during the binding-releasing process. The device may be either a MEMS device, such as quartz crystal 26 microbalance [1,2], or vibrating/bending cantilever [3], or 27 field effect transistor [4], or other [5]. The device with the 28 receptor molecules is called chemical sensor or detector. In 29 order to be useful, the detector must be able to discriminate 30 between different chemicals, to be selective. Its selectivity 31 is normally the same as that of its receptor molecules (see 32 33 Eqs. (6) and (7)).

The size of industrial sensors has constant tendency to de-34 35 crease [3]. The power of useful signal produced by a small detector becomes very small. As a result, noise of the detec-36 tor itself constitutes a substantial portion of its output signal. 37 Depending on its construction, there are several reasons for 38 a small detector to be noisy [6]. One type of noise is due to 39 the fact that the adsorption-desorption process is driven by 40

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Brownian motion, which is stochastic. As a result, the in-41 stantaneous total amount of adsorbed analyte is subjected to 42 irregular fluctuations visible in the output signal. This noise 43 is called the adsorption-desorption noise [7]. It is present in 44 any small detector which is based on binding-releasing of 45 analyte. The adsorption-desorption noise can dominate over 46 all other types of intrinsic noise [8]. 47

In this paper only the adsorption-desorption noise is taken into account. The detector is expected to be a threshold detector (ThD, Fig. 1).

Namely, the fluctuating signal characterizing the amount 51 of adsorbed analyte in the primary sensing unit (PSU in 52 Fig. 1) is fed into amplitude discriminator unit (threshold 53 unit, ThU in Fig. 1). The threshold unit is characterized by 54 a certain threshold. It has zero as its output if the adsorbed 55 amount is below the threshold, and it outputs standard dc sig-56 nal while the adsorbed amount is equal or above the thresh-57 old. The output of ThU is averaged over a sliding time win-58 dow to have final output practically time-independent. This 59 signal is considered as the output of the ThD. 60

In this paper, the temporal properties of the binding-61 releasing stochastic process are utilized to characterize the 62 outputs of ThD if two analytes are separately presented at 63 equal concentrations. This allows to compare selectivity of 64 ThD with that of its receptor molecules. The main conclu-65 sion is that the ThD may be much more selective than do 66 its adsorbing sites. 67

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Fig. 1. Schematic picture of ThD. A: analyte molecules; R: adsorption sites; PSU: primary sensing unit; ThU: threshold unit; TAU: temporal averaging unit.

68 2. Definitions and assumptions

The adsorption–desorption process is described by the following association–dissociation chemical reaction:

$$A + R \underset{k_{-}}{\overset{k_{+}}{\underset{k_{-}}{\underset{k_{-}}{\overset{k_{+}}{\underset{k_{-}}{\underset{k_{-}}{\overset{k_{+}}{\underset{k_{-}}{\underset{k_{-}}{\overset{k_{+}}{\underset{k_{-}}{\underset{k_{-}}{\underset{k_{-}}{\overset{k_{+}}{\underset{k_{-}}{k_{-}}{\underset{k_{-}}{k_{-}}{\atopk_{-}}{\underset{k_{-}}{$$

72 where A, R, and AR denote molecules of analyte, adsorption site or receptor, and analyte-receptor binary complex, 73 respectively. At constant temperature, the rate constants, 74 k_+, k_- are time-independent. They can be determined either 75 from experimental measurements, or estimated theoretically 76 [8]. Let N denote the total number of receptor molecules per 77 detector. The analyte is presented at concentration c. The 78 79 probability P for any R to be bound with A is¹

$$P = \frac{k_+ c}{k_+ c + k_-}.$$
 (2)

The mean number of adsorbed molecules, $\langle n \rangle$, can be calculated as follows:

83 $\langle n \rangle = PN.$

84 If two different analytes A_1 , A_2 are tested at the same con-85 centration, either Eq. (2), or experimental measurements will 86 give two values, P_1 , P_2 . We say that the receptor molecule 87 has selectivity with respect to A_1 , A_2 , if $P_1 \neq P_2$ (expect, 88 $P_1 > P_2$). The molecular selectivity, μ , is defined as², ³

$$\mu = \ln \frac{P_1}{P_2}.\tag{3}$$

The primary signal, $S_0(t)$ in Fig. 1, usually increases if the 90 number *n* of adsorbed molecules increases: 91

$$n > n' \Rightarrow S_0 > S'_0, \tag{4}$$

where the exact dependence of S_0 on *n* is determined by the sensor construction and the transduction mechanism it employs. For simplicity, it is expected that in the case of gravimetric sensor, A_1 and A_2 have equal molecular masses. Define selectivity δ for a whole detector in terms of final output signal (*S* in Fig. 1) as follows:

$$\delta = \ln \frac{S_1}{S_2},\tag{5}$$

where S_1 , S_2 are the final outputs for analytes A_1 , A_2 , respectively.

Both $S_0(t)$ and n(t) are subjected to adsorption–desorption 102 noise. In a detector without the threshold unit, the final output signal can be made linearly proportional to the mean 104 number of adsorbed molecules: 105

$$S_i \sim P_i N, \quad i = 1, 2.$$
 (6) 106

This is achieved either by temporal averaging, or choosing107large detector with powerful primary signal in which contri-108bution of adsorption-desorption fluctuations is not visible.109Substituting (6) into (5) one obtains for selectivity of a con-110ventional detector:111

$$\delta = \ln \frac{P_1 N}{P_2 N} = \mu.$$
 (7) 112

Thus, selectivity of detector in which the fluctuations are 113 averaged out either immediately after the primary sensing 114 unit, or inside it is equal to that of its individual adsorbing 115 sites. 116

The threshold unit, ThU, rises a threshold which the S_0 117 must overcome in order to make possible further stages of 118 processing. The crossing may happen from time to time 119 thanking to the adsorption–desorption fluctuations. Due to 120 (4), the threshold can be characterized by the number N_0 121 of analyte molecules which must be adsorbed before the 122 nonzero signal appears at the output end of the ThU. It is 123

¹ See [9], where Eq. (2) is justified.

² If one do not expect that $P_1 > P_2$ than Eq. (3) should be replaced by $\mu = |\ln(P_1/P_2)|$.

³ This definition of selectivity differs from used in chemistry the specificity of association which is expressed in terms of dissociation constant. For analyte A, the dissociation constant is defined as $[A]_{1/2} = k_-/k_+$. Eq. (2) can be rewritten using the dissociation constant: $P = 1/(1 + [A]_{1/2}/c)$. From this equation it is clear that analytes with different dissociation constants have different binding probabilities and vice versa. This proves suitability of both descriptions, even if numerical values of selectivity expressed in terms of dissociation constants, say as $\mu' = \ln([A_2]_{1/2}/[A_1]_{1/2})$, will differ from used here. The μ values can be expressed in terms of dissociation constants: $\mu = \ln((c + [A_2]_{1/2})/(c + [A_1]_{1/2}))$. The main difference between the μ' and μ is that the latter depends on concentration. This is in accordance with situation in natural olfactory systems where discriminating ability usually depends on concentration [10].

assumed that the ThU is ideal in a sense that the N_0 is the exact value which is not subjected to fluctuations. If N_0 is achieved, the ThU has standard constant signal as its output. The signal does not depend on the exact value of n(t) provided it is above or equal to N_0 .

Denote by *T* the temporal window over which the averaging is made in the TAU (Fig. 1), and by T_b , $T_a(T_b + T_a = T)$ the total amount of time during which n(t) is below or above the threshold, respectively, when $0 \le t \le T$. The final output, *S* in Fig. 1, should be linearly proportional to T_a/T . This gives for the selectivity of ThD:

$$\delta = \ln \frac{T_{a1}}{T_{a2}},\tag{8}$$

where T_{a1} and T_{a2} correspond to A₁ and A₂, respectively.

137 3. Estimation of selectivity

In accordance with (8), it is necessary to estimate the total amount of time the n(t) spends above the threshold when $t \in [0; T]$. This can be done by adding together lengths of all separate intervals during which $n(t) \ge N_0$ continuously. Denote by ζ the number of those intervals, and by T_a^k , $1 \le 144$ $k \le \zeta$ the length of the *k*th continuous interval. Then

$$T_a = \sum_{1 \le k \le \zeta} T_a^k = \zeta \frac{1}{\zeta} \sum_{1 \le k \le \zeta} T_a^k$$
$$= T \frac{(1/\zeta) \sum_{1 \le k \le \zeta} T_a^k}{(1/\zeta) \sum_{1 \le k \le \zeta} (T_b^k + T_a^k)}$$

147 where T_b^k is the length of *k*th continuous interval during 148 which $n(t) < N_0$. If *T* together with k_+ , *c*, k_- ensures that 149 ζ is large, then the last expression can be rewritten in the 150 following form:

$$T_a = T \frac{\bar{T}_{ac}}{\bar{T}_{bc} + \bar{T}_{ac}},\tag{9}$$

where T_{bc} , T_{ac} are the mean lengths of the continuous intervals. For the \overline{T}_{bc} , \overline{T}_{ac} the following expressions have been obtained [11] based on the Kolmogoroff (or backward Master) equation:

$$\bar{T}_{bc} = \frac{1}{k_{-}N_{0}C_{N}^{N_{0}}P^{N_{0}}(1-P)^{N-N_{0}}} \sum_{0 \le l < N_{0}} C_{N}^{l}P^{l}(1-P)^{N-l},$$
158
(10)

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$$\bar{T}_{ac} = \frac{1}{k_{-}N_{0}C_{N}^{N_{0}}P^{N_{0}}(1-P)^{N-N_{0}}} \sum_{N_{0} \le l \le N} C_{N}^{l}P^{l}(1-P)^{N-l}.$$
161
(11)

162 If two analytes, A_1 , A_2 are considered, then in (10) and 163 (11), k_- and P should be replaced with k_{-i} , P_i , i = 1, 2,

Table	1		
TT 1			

The rate constants used in the examples of Table 2 and in Fig. 2

	$k_+ (1/(s M))$	k_{-} (1/s)	
A ₁	1000	1000	
A ₂	1000	1050	

respectively. Substituting (10) and (11) into Eq. (9) one 164 obtains⁴

$$T_a = T \sum_{N_0 \le l \le N} C_N^l P^l (1 - P)^{N-l}.$$
 (12)

Considering (12) for two analytes, use it in Eq. (8). This 167 gives 168

$$\delta = \ln \frac{\sum_{N_0 \le l \le N} C_N^l P_1^l (1 - P_1)^{N-l}}{\sum_{N_0 \le l \le N} C_N^l P_2^l (1 - P_2)^{N-l}}.$$
(13)

The last equation can be replaced by a transparent estimate 170 if one use the following inequality: 172

$$\frac{\sum_{N_0 \le l \le N} C_N^l P_1^l (1 - P_1)^{N-l}}{\sum_{N_0 \le l \le N} C_N^l P_2^l (1 - P_2)^{N-l}} > \left(\frac{P_1}{P_2}\right)^{N((N_0/N - P_1)/(1 - P_1))},$$
(14) 174

which is proven in [11]. Substitution of (14) into Eq. (13) 175 gives 176

$$\delta > N \frac{P_0 - P_1}{1 - P_1} \mu, \quad P_0 = \frac{N_0}{N}.$$
 (15) 177

Taking into account that the total number of adsorbing sites, 178 N, as well as N_0 can be very large, it is clear from the 179 estimate (15) that δ can be much larger than μ , provided the 180 fraction $(P_0 - P_1)/(1 - P_1)$ is not very small. It must be at 181 least positive, which requires 182

$$P_0 > P_1 \quad \text{or} \quad P_1 N < N_0.$$
 (16) 183

Taking into account that P_1 increases with concentration 184 (see Eq. (2)), inequality (16) can be considered as impos-185 ing an upper limit for concentration c at which the effect 186 of selectivity improvement might be expected based on the 187 estimate (15). It is worth to notice that when condition (16)188 holds, the mean amount of adsorbed analyte is below the 189 threshold one, and threshold crossing may happen only due 190 to fluctuations. 191

4. Numerical examples

As one can conclude from the estimate (15), the selectivity improvement is higher for higher N_0 (Table 1). On the other hand, one cannot choose the N_0 as high as desired because the ThU in Fig. 1 is expected to be ideal. If one chose

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⁴ The following relation is used: $\sum_{0 \le l \le N} C_N^l P^l (1-P)^{N-l} = 1$. See also [9], where equivalent to (12) conclusion is obtained based on simpler and less rigorous reasoning not using Eqs. (10) and (11).



Fig. 2. Concentration dependencies of selectivity for the examples of Table 2. Concentration (*x*-axis) is given in M. The N, N_0 values in a and b correspond to the first and second rows of Table 2, respectively. The δ_{est} corresponds to the right hand side of the inequality (15).



Fig. 3. Short segment of the trajectory n(t) modeled on PC for the Example 1 of Table 2. Time (x-axis) is given in seconds.

 $N_0 = 100$ then the ideality means that the threshold level in 197 the ThU is allowed to have less then 1% jitter. Similarly, if 198 one chose $N_0 = 10^4$ then the threshold level must be kept 199 with better than 0.01% precision. Otherwise, noise in the 200 threshold level should be taken into account in the reasoning 20 of n. 3, and this will lead to a less promising estimate. An-202 other conclusion, based on the estimate (15), suggests that 203 the smaller is the concentration (smaller P_1) of the analytes, 204 205 the better is discrimination between them. But in this case the threshold will be achieved during small fraction of time 206 spent for measuring. As a result, the output signal will be 207 very small and may be lost in the TAU unit. It is natural 208 209 to require that the output signal for more affine analyte is higher than the 10% of the maximal output signal, which 210 is produced if $n(t) \ge N_0$ all the time. Taking into account 211 Eq. (12) this leads to the following constraint: 212

$$r_1 = \sum_{N_0 < l < N} C_N^l P_1^l (1 - P_1)^{N-l} > 0.1.$$
(17)

One more constraint comes from assumption of large ζ 214 215 which is made for derivation of Eq. (9). If the measuring (averaging) time T is to be short enough, say T = 1 s, then 216 the mean frequency of crossing the threshold should be high 217 enough in order to have, e.g., $\zeta > 1000$. This could be 218 achieved if the mean durations of being continuously above 219 and below the threshold are short enough. If $\overline{T}_{abc} = \overline{T}_{bc} + \overline{T}_{ac}$, 220 then Eqs. (10) and (11) give 221

222
$$\bar{T}_{abc} = (k_{-}N_0C_N^{N_0}P^{N_0}(1-P)^{N-N_0})^{-1}$$

223 The $\zeta > 1000$ could be ensured by the following inequality

$$\frac{T}{\bar{T}_{abc}} > 1000,$$

225 or, choosing T = 1 s:

226
$$\bar{T}_{abc} < 0.001 \text{ s.}$$

 Table 2

 Numerical examples of improved selectivity^a

	N	N_0	c (M)	μ	δ	r_1	\bar{T}_{abc1} (s)
Example 1	10 ⁷	10^{3}	9.6×10^{-5}	0.05	3.63	0.1	1.8×10^{-4}
Example 2	10 ⁸	10^{4}	9.9×10^{-5}	0.05	18	0.16	4×10^{-5}

^a The rate constants for the analytes are shown in Table 1. δ is calculated here by means of the exact expression (13), r_1 —as shown in Eq. (17).

Two examples satisfying this constraints are shown in 227 Table 2. Concentration dependencies of μ , δ , and the estimate (15) are shown in Fig. 2. A short segment of the 229 trajectory n(t) modeled on PC is shown in Fig. 3. 230

5. Conclusions

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In this paper, selectivity of chemical sensor is compared 232 with that of its primary receptors (adsorbing sites). The sen-233 sor is expected to be a small one, in which the main source 234 of noise is due to the adsorption-desorption fluctuations. In 235 the sensor considered, the signal from the primary sensing 236 unit is immediately subjected to the amplitude discrimina-237 tion defined in Section 1, and obtained piecewise-constant 238 signal (L(t) in Fig. 1) is averaged over a time window. The 239 averaged signal (S in Fig. 1) is taken as the output of whole 240 sensor. 241

The threshold-crossing statistics derived from the exact 242 description of the adsorption–desorption stochastic process 243 is used for estimating selectivity. As a result, it is concluded 244 that selectivity of this sensor can be much better than that 245 of its primary receptors. The effect may be expected in a 246 limited range of concentrations of analytes, which depends 247 on the threshold level. For high concentrations the selectiv- 248

ity falls to that of the primary receptors (Fig. 2), and for
low ones the output signal will be too small even for more
affine analyte. The best situation is expected when the mean
number of bound receptors is just below the threshold one,
and the threshold is frequently crossed due to the presence
of fluctuations. Thus, in practical realization a possibility of
tunable threshold should be considered.

256 6. Discussion

Usually, noise in sensory devices is taken as unfavorable 257 factor.⁵ In this consideration, the presence of noise looks 258 like factor improving the sensor's performance. But with the 259 ideal threshold unit in hands much can be done even without 260 noise. Expect that the noise is initially averaged out either 261 262 by spatial averaging (choosing big primary unit with large N), or by temporal averaging (interchanging TAU with ThU 263 in Fig. 1). The averaged signals for the A_1 , A_2 can be very 264 close (see Eq.??), but the ideal ThU with tunable thresh-265 266 old will be able to discriminate perfectly between them. Thus, even if the fluctuations in this sensor are made work-267 ing, the answer what is better to do first for the practical 268 purposes: the amplitude discrimination, or temporal averag-269 ing, depends on physical parameters of the environment in 270 which the sensor operates, and on physical characteristics of 271 the sensor itself, including intensity of noises other than the 272 273 adsorption-desorption one. Interesting, in natural olfactory systems, a kind of amplitude discrimination is made imme-274 diately after the primary reception [11,13]. Also in those 275 systems the threshold is tunable due to adaptation of indi-276 vidual neurons. 277

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Biography

Alexander K. Vidybida received his PhD in Mathematical and Theo-318 retical Physics (Kandidat of Physics and Mathematics) in 1975 at the 319 Institute for Theoretical Physics, Kyiv, for his study of the BBGKY hier-320 archy of equations in nonequilibrium statistical mechanics. He received 321 his DrHab degree in Theoretical Physics (Doktor of Physics and Mathe-322 matics) in 2000 at the Institute for Theoretical Physics, Kyiv, for his study 323 of interaction of alternating electromagnetic fields with macromolecular 324 and cooperative systems, including interaction of microwaves with liv-325 ing objects. At present time he works as Senior Scientist in the Institute 326 for Theoretical Physics. Dr. A.K. Vidybida is a Member of Euroscience, 327 IBRO (International Brain Research Organization), EBEA (European Bio-328 electromagnetic Association). His current research interests are devoted to 329 neurophysics, including formation of high discriminating ability in natural 330 sensory systems, as well as to implementation of basic ideas in technical 331 devices. 332

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⁵ But see [12], where some characteristics of noise are employed for discriminating purposes.